What exactly is fire? I know it's combustion of fuel, blah, blah... but what is it exactly? Is it purely energy? What state of matter is it? I suspect it's highly energetic gases, whose energy state is so high that they emit light, and thus we see flame, with the hotter flames being higher in electromagnetic energy. But I've never seen any source explaining exactly what fire is. Can it be ionized? Can it be affected by magnetism? Gravity?

—Wenhsingyu

his letter shows why the world needs me. The explanations of fire in most standard reference works (not to put too fine a point on it) suck. The Encyclopaedia Britannica, for

example, takes the following ineffectual stab: "rapid burning of combustible material with the evolution of heat and usually accompanied by flame." No wonder people have lost faith in science. Time to let a professional show how it's done. We'll start by reviewing the defects of

the *EB* definition.

(1) *Rapid burning*. OK, fire burns rapidly. Slow combustion, also known as oxidation, includes such nonfiery processes as rusting and digestion. However, to say that fire involves burning is to embark on a circular definition: What is fire? When something burns. What is burning? When something is on fire. We need to address the matter in more basic terms.

(2) Combustible material. Redundant. If a material is capable of rapid burning, by definition it's combustible. Also, "material" is needlessly vague. One specific material, oxygen, is always involved.

(3) Evolution of heat. I don't know what this is

supposed to mean either.

(4) Usually accompanied by flame. "Usually accompanied by" my arse. Flame is the quintessential pyrolytic phenomenon, not a mere by-product.

Now for a proper definition, devised by myself: "Fire is the rapid combination of oxygen with fuel in the presence of heat, typically characterized by flame, a body of incandescent gas that contains and sustains the reaction and emits light and heat." Let's go through this bit by bit:

(1) Rapid combination of oxygen with fuel in the presence of heat. Oxygen, fuel, and heat are the es-

sential ingredients of fire.

(2) Typically characterized by flame. The pup qualifier "typically" allows me to sidestep the issue of apparently nonflaming fires, like you get with burning charcoal. I suspect charcoal fires do create flame—you just can't see it due to the lack of impurities or incompletely burned fuel in the plume. (You can't see a fuel fire at a NASCAR race either, because the cars run on clean-burning methanol.) But that's a matter we can leave for another day.

(3) Body of incandescent gas. Flame defined. Most

encyclopedia and dictionary definitions blow past this entirely, allowing persons such as yourself to imagine that fire is "pure energy" or similar nonsense. We say "body" because the gas has a characteristic structure and composition. We say "incandescent" because (a) it sounds scientific, (b) it means "luminous with intense heat," precisely what we are attempting to convey, and (c) if the Teeming Millions are going to learn one vocabulary word today, by God "incandescent" should be it.

(4) Contains and sustains the reaction. Flame isn't just the result of fire; it is the fire. What's more, without the flame's heat the fire



(5) Emits light and heat. Duh. However, we mustn't overlook the obvious. Now to your other questions:

• Is fire purely energy? Clearly not. The dancing flames are glowing gas—your suspicions confirmed.

• Can fire be ionized? Is it affected by magnetism? Not so's you'd notice. You're thinking of plasma, which is ionized (and thus electromagnetically reactive) gas, often described as the fourth state of matter. You see it in welding arcs, lightning bolts, and the sun. Ordinary fire isn't plasma.

• Is fire affected by gravity? Of course—gas has mass. Flame is shaped by convection, a function of gravity (hot air rises). In low- or zero-G environments, fire looks way different: a candle flame on the space shuttle isn't yellow and tapered but blue and nearly spherical (see www3.cosmiverse.com/news/space/0802/space08220205.html).

So there you have it. One step closer to a better world.